

# Semi-quantitative LC/ESI/MS analysis using predictive models of ESI ionization efficiencies

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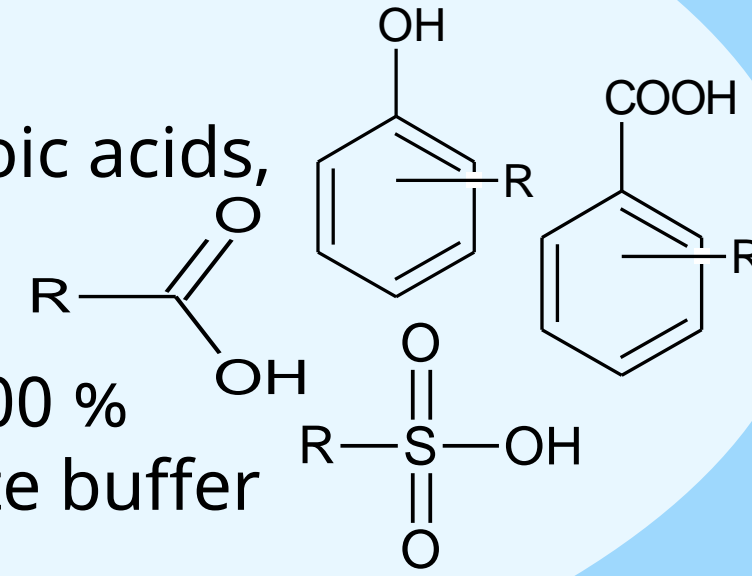
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## Studied Compounds and Eluents

### Training set:

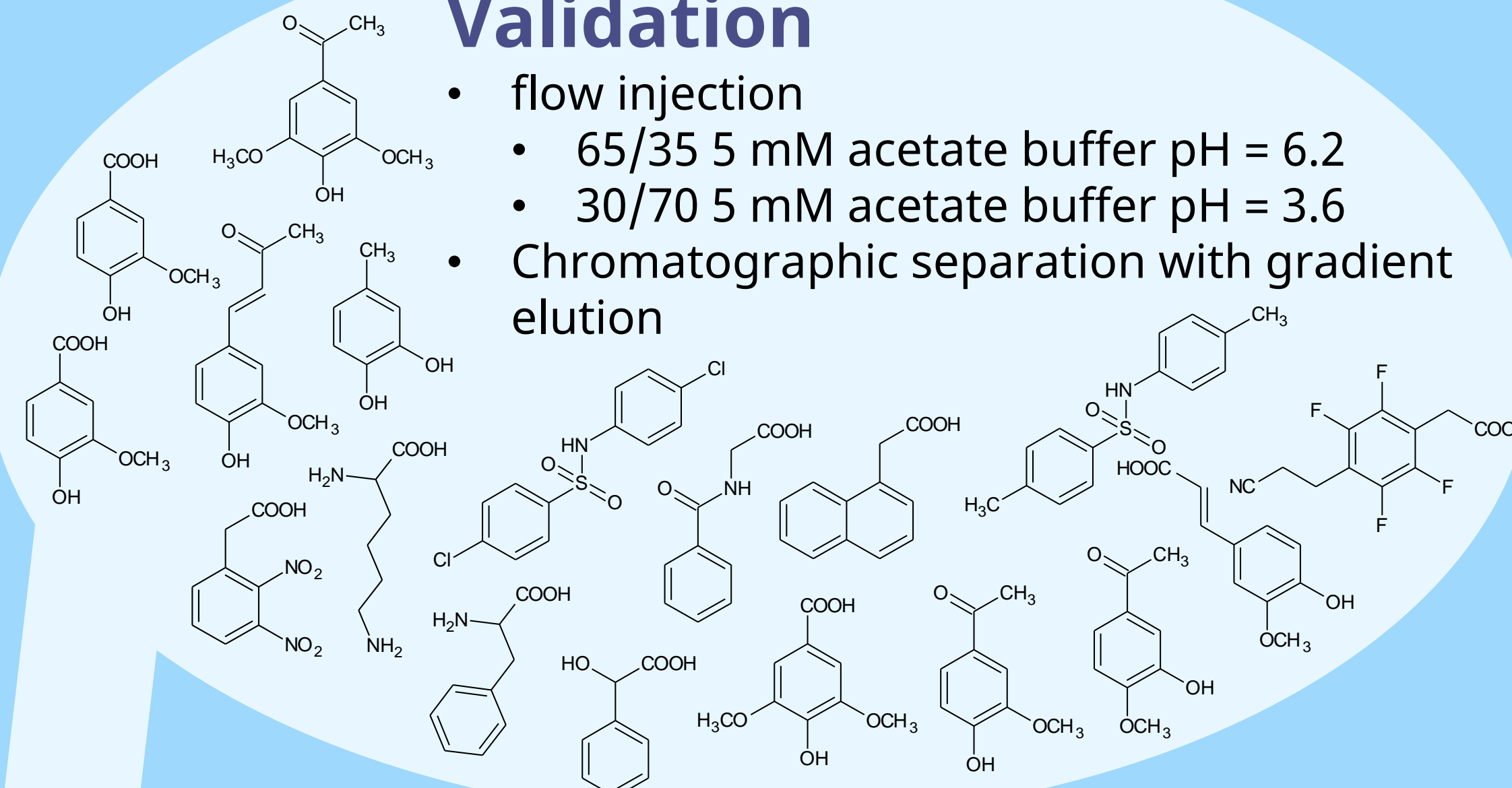
- 62 substituted phenols and benzoic acids, carboxylic and sulfonic acids
- 10 eluent compositions
- Acetonitrile percentage 20 - 100 %
- Additives: NH<sub>3</sub>, HCOOH, acetate buffer
- pH = 2.7 - 10.7



## Validation

### flow injection

- 65/35 5 mM acetate buffer pH = 6.2
- 30/70 5 mM acetate buffer pH = 3.6
- Chromatographic separation with gradient elution



## Electrospray Ionization Efficiency (IE)

- shows **how efficiently** analyte ions from liquid phase form gas phase ions
- allows to study **processes** occurring during **ESI plume** and in **droplet**, giving more insight into **ESI mechanism**
- depends on **physico-chemical properties** of **analyte** and **eluent**
- Relating analyte and eluent physico-chemical parameters to **IE** gives the power to predict **IE**
- Ability to **predict IE** allows to carry out **standard substance free quantification** with any needed analyte in any preferable medium

## Descriptors

### PaDEL

- 1444 1D and 2D
- 431 3D
- freeware
- quick and easy

- hard to explain chemical meaning
- trivial descriptors (e.g. nN)
- uses gas phase geometry

### COSMO-RS

- descriptors have chemical meaning
- Possible to calculate for solvent mixtures

- 20 3D descriptors
- time-consuming geometry optimization
- licensed software

## Model Development

$$\log IE = (2.72 \pm 0.13) + (-0.46 \pm 0.02) \cdot WAPS + (0.69 \pm 0.07) \cdot \alpha + (-0.018 \pm 0.004) \cdot HBA + (0.0039 \pm 0.0011) \cdot \%MeCN$$

WAPS - charge delocalization parameter,  $\alpha$  - ionization degree in solution, HBA - hydrogen bond acceptor ability, %MeCN - acetonitrile content

## Calibrating the Setup of Interest

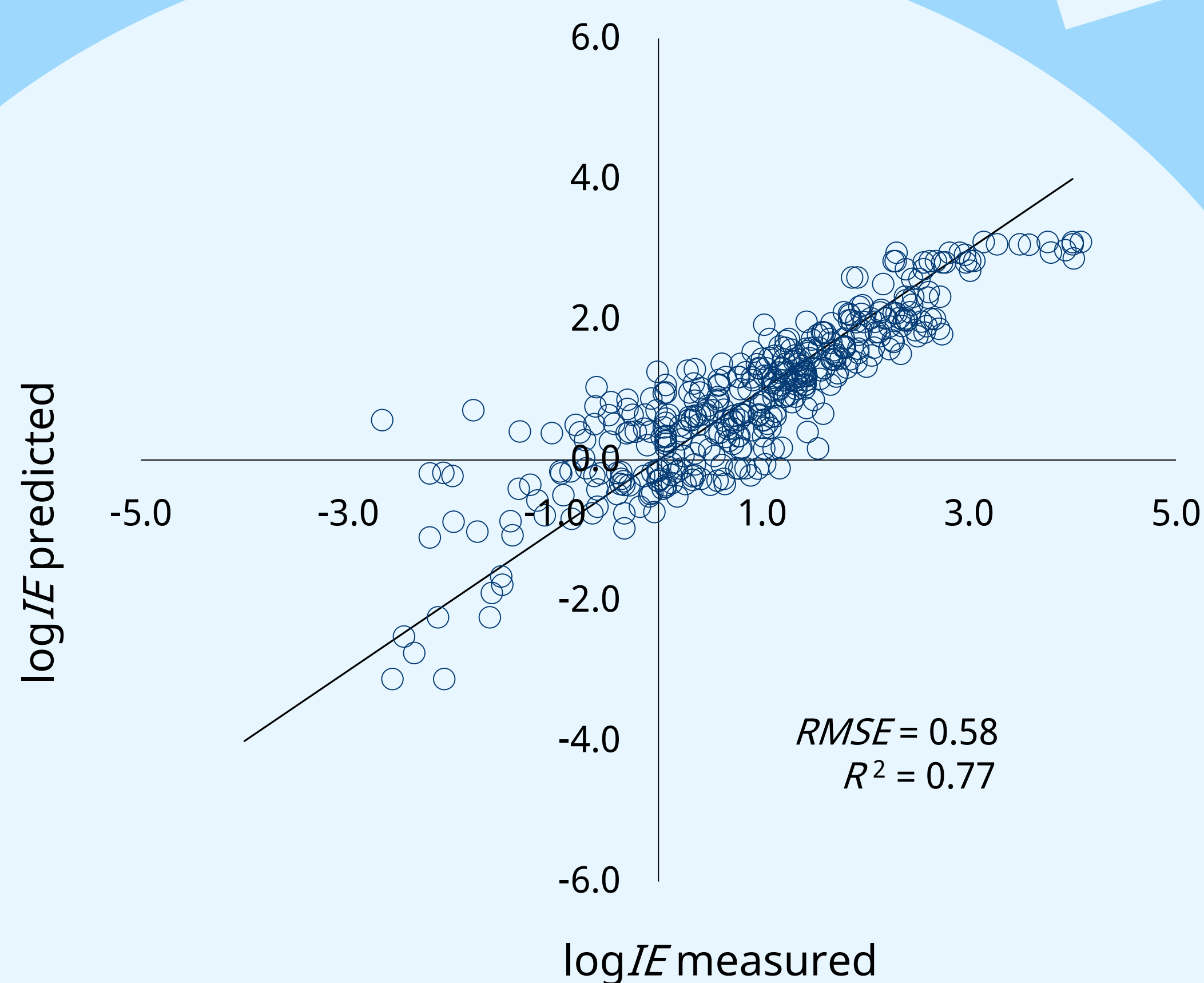
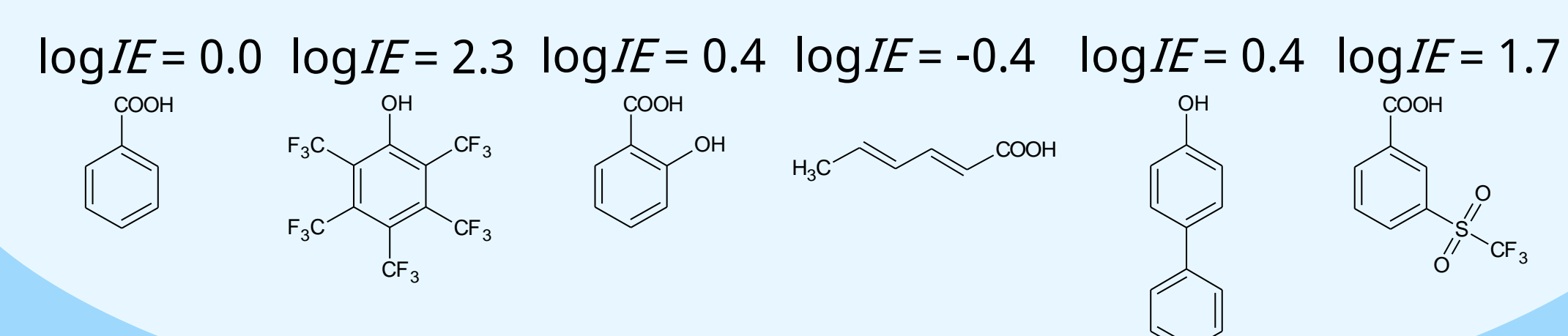


Figure 1: Fit obtained for the log IE values over all tested mobile phases. Straight line shows ideal fit.

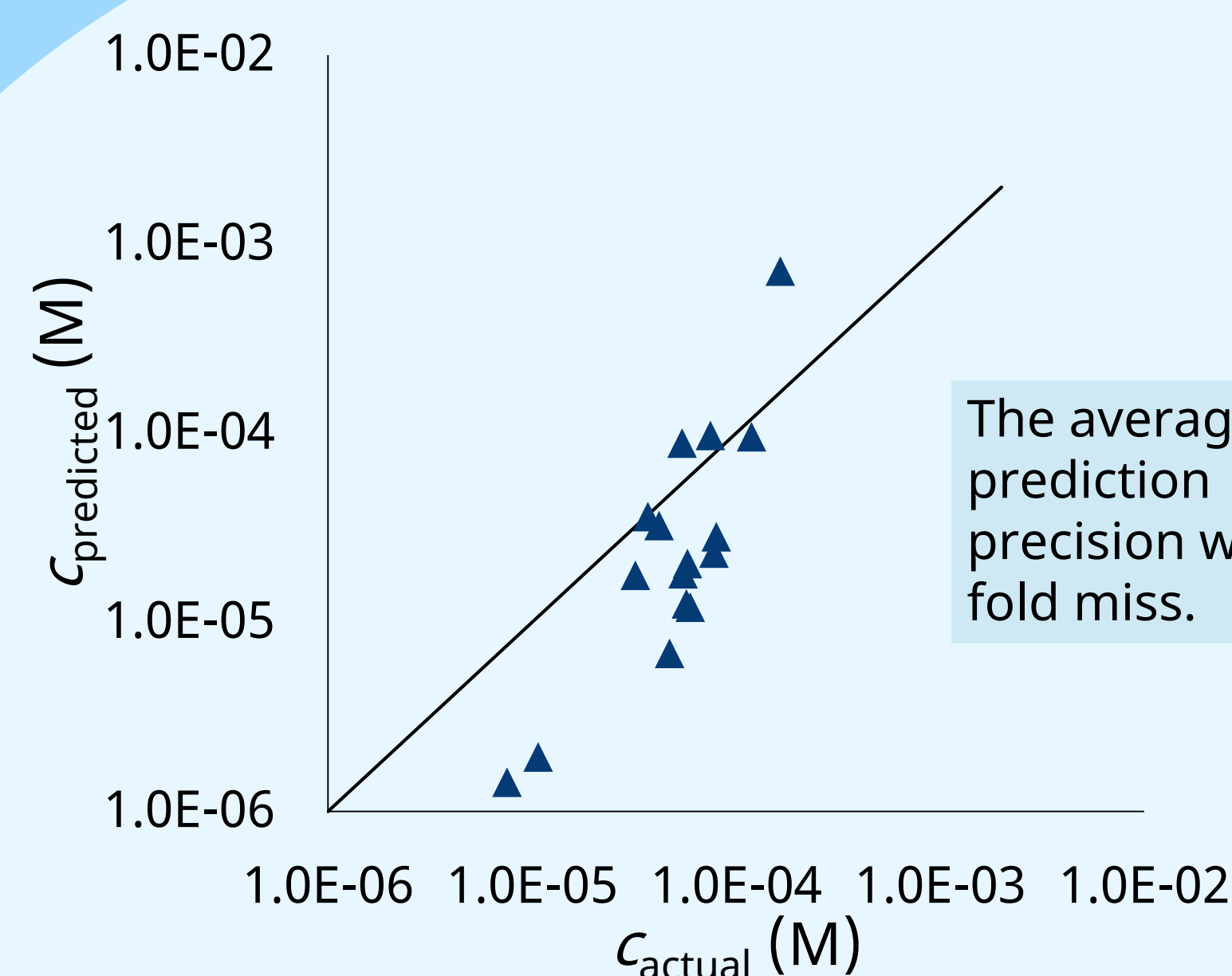


Figure 2: Correlation between the predicted concentrations and measured concentrations for the 16 acids in chromatographic mode.

## References

- Kruve and Kaupmees *Anal Chem* 89(9), 2017
- Liigand et al *Anal Chem* 89(11), 2017
- Liigand et al *J Am Soc Mass Spectrom* 28(3), 2017
- Kruve et al *J Am Soc Mass Spectrom* 28(5), 2017

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